



UNITED STATES PATENT AND TRADEMARK OFFICE

UNITED STATES DEPARTMENT OF COMMERCE
United States Patent and Trademark Office
Address: COMMISSIONER FOR PATENTS
P.O. Box 1450
Alexandria, Virginia 22313-1450
www.uspto.gov

APPLICATION NO.	FILING DATE	FIRST NAMED INVENTOR	ATTORNEY DOCKET NO.	CONFIRMATION NO.
-----------------	-------------	----------------------	---------------------	------------------

10/591,523

09/01/2006

Yoshiyuki Asahina

10993.0282

2959

22852

7590

03/17/2010

FINNEGAN, HENDERSON, FARABOW, GARRETT & DUNNER
LLP

901 NEW YORK AVENUE, NW
WASHINGTON, DC 20001-4413

EXAMINER

SERGEANT, RABON A

ART UNIT

PAPER NUMBER

1796

MAIL DATE

DELIVERY MODE

03/17/2010

PAPER

Please find below and/or attached an Office communication concerning this application or proceeding.

The time period for reply, if any, is set in the attached communication.

Office Action Summary	Application No. 10/591,523	Applicant(s) ASAHINA ET AL.	
	Examiner Rabon Sergent	Art Unit 1796	

-- The MAILING DATE of this communication appears on the cover sheet with the correspondence address --

Period for Reply

A SHORTENED STATUTORY PERIOD FOR REPLY IS SET TO EXPIRE 3 MONTH(S) OR THIRTY (30) DAYS, WHICHEVER IS LONGER, FROM THE MAILING DATE OF THIS COMMUNICATION.

- Extensions of time may be available under the provisions of 37 CFR 1.136(a). In no event, however, may a reply be timely filed after SIX (6) MONTHS from the mailing date of this communication.
- If NO period for reply is specified above, the maximum statutory period will apply and will expire SIX (6) MONTHS from the mailing date of this communication.
- Failure to reply within the set or extended period for reply will, by statute, cause the application to become ABANDONED (35 U.S.C. § 133). Any reply received by the Office later than three months after the mailing date of this communication, even if timely filed, may reduce any earned patent term adjustment. See 37 CFR 1.704(b).

Status

- 1) ☒ Responsive to communication(s) filed on 23 December 2009.
- 2a) ☒ This action is **FINAL**. 2b) ☐ This action is non-final.
- 3) ☐ Since this application is in condition for allowance except for formal matters, prosecution as to the merits is closed in accordance with the practice under *Ex parte Quayle*, 1935 C.D. 11, 453 O.G. 213.

Disposition of Claims

- 4) ☒ Claim(s) 1-13 is/are pending in the application.
- 4a) Of the above claim(s) _____ is/are withdrawn from consideration.
- 5) ☐ Claim(s) _____ is/are allowed.
- 6) ☒ Claim(s) 1-13 is/are rejected.
- 7) ☐ Claim(s) _____ is/are objected to.
- 8) ☐ Claim(s) _____ are subject to restriction and/or election requirement.

Application Papers

- 9) ☐ The specification is objected to by the Examiner.
- 10) ☐ The drawing(s) filed on _____ is/are: a) ☐ accepted or b) ☐ objected to by the Examiner.
Applicant may not request that any objection to the drawing(s) be held in abeyance. See 37 CFR 1.85(a).
Replacement drawing sheet(s) including the correction is required if the drawing(s) is objected to. See 37 CFR 1.121(d).
- 11) ☐ The oath or declaration is objected to by the Examiner. Note the attached Office Action or form PTO-152.

Priority under 35 U.S.C. § 119

- 12) ☒ Acknowledgment is made of a claim for foreign priority under 35 U.S.C. § 119(a)-(d) or (f).
- a) ☒ All b) ☐ Some * c) ☐ None of:
1. ☐ Certified copies of the priority documents have been received.
 2. ☐ Certified copies of the priority documents have been received in Application No. _____.
 3. ☒ Copies of the certified copies of the priority documents have been received in this National Stage application from the International Bureau (PCT Rule 17.2(a)).

* See the attached detailed Office action for a list of the certified copies not received.

Attachment(s)

- | | |
|---|---|
| 1) <input type="checkbox"/> Notice of References Cited (PTO-892) | 4) <input type="checkbox"/> Interview Summary (PTO-413) |
| 2) <input type="checkbox"/> Notice of Draftperson's Patent Drawing Review (PTO-948) | Paper No(s)/Mail Date. _____ |
| 3) <input type="checkbox"/> Information Disclosure Statement(s) (PTO/SB/08) | 5) <input type="checkbox"/> Notice of Informal Patent Application |
| Paper No(s)/Mail Date _____ | 6) <input type="checkbox"/> Other: _____ |

Art Unit: 1796

1. Claim 6 is rejected under 35 U.S.C. 112, second paragraph, as being indefinite for failing to particularly point out and distinctly claim the subject matter which applicant regards as the invention.

In view of the amendment to claim 1, the limitation of claim 6 fails to be further limiting.

2. The following is a quotation of 35 U.S.C. 103(a) which forms the basis for all obviousness rejections set forth in this Office action:

(a) A patent may not be obtained though the invention is not identically disclosed or described as set forth in section 102 of this title, if the differences between the subject matter sought to be patented and the prior art are such that the subject matter as a whole would have been obvious at the time the invention was made to a person having ordinary skill in the art to which said subject matter pertains. Patentability shall not be negated by the manner in which the invention was made.

This application currently names joint inventors. In considering patentability of the claims under 35 U.S.C. 103(a), the examiner presumes that the subject matter of the various claims was commonly owned at the time any inventions covered therein were made absent any evidence to the contrary. Applicant is advised of the obligation under 37 CFR 1.56 to point out the inventor and invention dates of each claim that was not commonly owned at the time a later invention was made in order for the examiner to consider the applicability of 35 U.S.C. 103(c) and potential 35 U.S.C. 102(e), (f) or (g) prior art under 35 U.S.C. 103(a).

3. Claims 1-13 are rejected under 35 U.S.C. 103(a) as being unpatentable over Asahina et al. ('048) in view of Slack et al. ('519) or Bernard et al. (US 2003/0096909 A1).

Asahina et al. disclose polyisocyanates, blocked polyisocyanates, and coating compositions derived from reacting the polyisocyanate or blocked polyisocyanate with a polyol, wherein the polyisocyanates have a number of terminal isocyanate groups of 5 to 20 and are produced by reacting an aliphatic diisocyanate, cycloaliphatic diisocyanate, or mixture thereof with a polyhydroxyl compound having a functionality of 4.5 to 10 at an NCO/OH ratio of 2:1 to

Art Unit: 1796

30:1 and removing substantially all unreacted diisocyanate monomer. Contrary to applicants' arguments within the response of December 23, 2009, the reference discloses suitable polyhydroxyl compounds having molecular weights below 300. See, for example, the disclosed polyol species; sorbitol, mannitol, and glucose; within column 8, lines 30+. Patentees further disclose the presence of cyclic trimer (isocyanurate) and allophanate groups and further disclose that the polyisocyanate has a viscosity as high as 200,000 mPa's/25°C. See abstract; column 6, lines 13+; column 7, lines 58+; column 8; column 10, lines 18+; column 11, lines 16+; column 12, lines 21; and column 16, lines 30-41. Furthermore, in view of the disclosure and examples, applicants' claimed polyol component concentration is considered to be met. See Table 1.

4. Though Asahina et al. disclose that mixtures of polyisocyanates can be used, patentees fail to disclose applicants' claimed ratio of aliphatic diisocyanate to alicyclic diisocyanate and further fail to disclose the claimed glass transition temperature range. However, the use of blends of aliphatic diisocyanate, such as hexamethylene diisocyanate, and alicyclic diisocyanate, such as isophorone diisocyanate, in the production of polyisocyanate compositions, intended for use in coating compositions was known at the time of invention. Slack et al. disclose such blends in the production of high viscosity, high equivalent weight polyisocyanate mixtures containing both allophanate and isocyanurate group and the use of the polyisocyanates to produce coatings. See abstract. Bernard et al. disclose within paragraph [0018] that blends of hexamethylene diisocyanates derivatives and isophorone diisocyanate derivatives are used to control the rate of surface drying and further disclose a relationship between surface hardness and glass transition temperatures. Therefore, the position is taken that that the teachings of the references establish that the ratio of aliphatic diisocyanate to alicyclic diisocyanate and the glass

Art Unit: 1796

transition temperatures of the polyisocyanates are result effective variables, and the position is taken that discovering optimum values of these result effective variables involves only routine skill in the art. *In re Boesch*, 617 F.2d 272, 205 USPQ 215 (CCPA 1980). Accordingly, the position is ultimately taken that it would have been obvious to arrive at the claimed ratios and glass transition temperatures, based on the teachings of the references.

5. Applicants' response has been considered; however, the response is insufficient to overcome the prior art rejection. Applicants' argument with respect to the molecular weight of the polyhydroxyl compound has been addressed above within paragraph 3. The examiner has considered applicants' arguments with respect to the secondary references; however, the position is maintained that these references serve to teach the desirability of using blends comprising aliphatic and alicyclic structures derived from polyisocyanates, and the presence of other reactants or structures within these blends is not considered to derogate the significance of these teachings. Within the confines of each secondary reference, one would expect that the relied upon teachings concerning the ratio of aliphatic to alicyclic structures would hold regardless of the presence of other reactants or structures, since these other reactants or structures can be considered to be constants or non-variables, within the context of each disclosure. In other words, where isocyanurate groups are present within all compositions yet the point of variance stems from the ratio of aliphatic to alicyclic groups, one would simply focus on the point of variance to ascertain any trends or effects. This rationale is analogous to the accepted practice of holding prior art variables constant except for those that are argued to yield the unexpected result. The examiner has considered applicants Example 8 and Comparative Example 9; however, the fact that a higher glass transition temperature is obtained for the example of the

Art Unit: 1796

invention is not considered to be meaningful, because the respective compositions are so different. This data doesn't rebut the examiner's position that it was known to employ the claimed structures in certain combinations, rather the data simply establishes that divergent compositions, each considered on the whole, possess different properties.

6. Applicant's amendment necessitated the new ground(s) of rejection presented in this Office action. Accordingly, **THIS ACTION IS MADE FINAL**. See MPEP § 706.07(a). Applicant is reminded of the extension of time policy as set forth in 37 CFR 1.136(a).

A shortened statutory period for reply to this final action is set to expire THREE MONTHS from the mailing date of this action. In the event a first reply is filed within TWO MONTHS of the mailing date of this final action and the advisory action is not mailed until after the end of the THREE-MONTH shortened statutory period, then the shortened statutory period will expire on the date the advisory action is mailed, and any extension fee pursuant to 37 CFR 1.136(a) will be calculated from the mailing date of the advisory action. In no event, however, will the statutory period for reply expire later than SIX MONTHS from the date of this final action.

Any inquiry concerning this communication should be directed to R. Sergent at telephone number (571) 272-1079.

/Rabon Sergent/
Primary Examiner, Art Unit 1796